

# Download Basicity Of Sulphuric Acid

I think , the basicity of sulphuric acid is 1. The basicity of an acid is equivalent to the number of  $H^+$  produced by an acid, when it gets ionized in the aqueous medium. So, for sulphuric acid, it is 2. One can say that the basicity of the acids is equal to the hydrogen atom present in the acid. In the last post, we have covered Chemical Definitions of ACIDS. Today we are going to cover on the “Basicity of ACIDS” – this part is where lots of students have misconception in their schools. The basicity of an acid is equivalent to the number of  $H^+$  produced by an acid, when it gets ionized in the aqueous medium. So, for sulphuric acid, it is 2. One can say that the basicity of the acids is equal to the hydrogen atom present in the acid.