

Download P6 Molecular Geometry

I just finished that question. The electron domain geometry for PF_6^- is octahedral and the molecular geometry is octahedral. The electron domain geometry for NH_3Cl^+ is tetrahedral and the molecular geometry is tetrahedral. I hate Mastering Chem, so I hope this helps. The electronic geometry gives the rough starting shape of the molecule but once the electron configuration is added it changes the shape of the molecule from the original electronic geometry... It helped me a lot. ^^ electron pair geometry and molecular geometry won't be the same if there are lone pairs involved.