

# Download Xef Compound

XeF<sub>2</sub> can act as a ligand in coordination complexes of metals. For example, in HF solution: Mg(AsF<sub>6</sub>)<sub>2</sub> + 4 XeF<sub>2</sub> → [Mg(XeF<sub>2</sub>)<sub>4</sub>](AsF<sub>6</sub>)<sub>2</sub>. Crystallographic analysis shows that the magnesium atom is coordinated to 6 fluorine atoms. Four of the fluorines are attributed to the four xenon difluoride ligands while the other two are a pair of cis-AsF<sub>6</sub> ligands. View Javadoc. 1 package org.apache.maven.plugin; 2 3 /\* 4 \* Licensed to the Apache Software Foundation (ASF) under one 5 \* or more contributor license agreements. This page was automatically generated by MavenThe U.S. Department of Energy's Office of Scientific and Technical Information